

Total Synthesis of Des-AB-cholestane-8 β ,9 α -diol

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Summary A brief and highly stereoselective total synthesis gives the title compound (7; R = H), from which 9 α -chloro-des-AB-cholestan-8 β -ol, used for the synthesis of precalciferol₃ can be prepared.

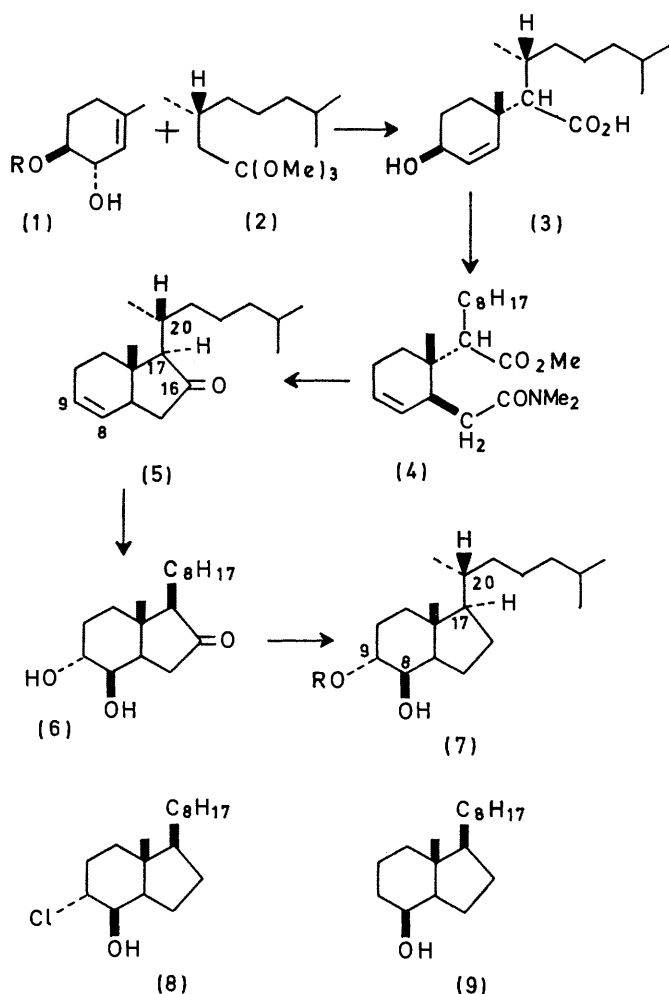
9 α -CHLORO-DES-AB-CHOLESTAN-8 β -OL (8),[†] used for the synthesis¹ of precalciferol₃ is a product of total synthesis by virtue of its preparation from des-AB-cholestan-8 β -ol (9).[†] The latter has been obtained² *via* relay compounds from 4-methoxycarbonyl-3-methylcyclohex-2-en-1-one, although only in very small (*ca.* 0.001%) yield. With the aim of providing a more efficient start to the synthesis of precalciferol₃, we have devised a brief and stereoselective total

synthesis of the title diol (7; R = H), from which the chlorohydrin (8) is easily prepared.

The synthesis takes its form and starting materials from the method used to secure the 20 β -configuration.³ We noted that the readily available (*R*)-dihydrocitronellic acid contains configurational and structural features which fit it to provide the iso-octyl side-chain together with C-17 and C-16 of the ketone (5). Model experiments showed that the orthoacetate variant⁴ of the Claisen rearrangement, hitherto used for the transfer of the $\cdot\text{CH}_2\cdot\text{CO}_2\text{Et}$ group in acyclic systems, could be extended to the transfer of $\cdot\text{CHR}\cdot\text{CO}_2\text{Et}$ groups in cyclohex-2-en-1-ols; this suggested the use of the orthoester (2). To unite it with a ring-c

[†] All the structures in this communication denote absolute configurations.

fragment, the methods outlined in the preceding communication⁵ were developed; they allow the assembly of all



eighteen carbon atoms of the ketone (5) in two synthetic steps, with the absolute configurations of three of its four asymmetric centres predetermined by those of the starting materials.

The orthoester (2) was obtained by standard methods from the dihydro-compound, $[\alpha]_D^{23} -4.6^\circ$ (neat), of citronellonitrile M.⁶ Reaction of the orthoester (3 mols) with the mesitoate⁷ (1; R = Me₃C₆H₂CO), followed by vigorous alkaline hydrolysis of the product, gave the hydroxy-acid (3); its methyl ester was then converted into the dimethylamide (4), which was hydrolysed to the corresponding dibasic acid. Cyclisation as previously described of the dimethyl ester gave a (near-equilibrium) mixture (ca. 6 : 1 by g.l.c.) of the crystalline (low-melting) hydronone (5), $[\alpha]_D^{21} -93^\circ$ (chloroform), and its 17 α -epimer. The 17 β -epimer (5) formed an 8 α ,9 α -epoxide m.p. 83°, reaction of which with dilute sulphuric acid in acetone gave the 8 β ,9 α -diol (6), m.p. 113–114°, $[\alpha]_D^{22} -122^\circ$ (chloroform). After acetylation and reaction with ethane dithiol and boron trifluoride etherate, the thioacetal was desulphurised with Raney nickel; deacetylation then gave the diol (7; R = H), m.p. 142–143°, $[\alpha]_D^{22} +49^\circ$ (chloroform), identical with material prepared⁸ from des-AB-cholest-8-ene.^{1,9} In these preliminary experiments the diol (7; R = H) was obtained in 13% yield from the mesitoate (1; R = Me₃C₆H₂CO). We expect that the use of other esters of the diol (1; R = H), may result in improvements; they will be reported in the full paper.

Selective monotosylation of the diol (7; R = H) gives⁹ in good yield the monotosy late (7; R = MeC₆H₄SO₂), which is converted by standard methods into the 8 β ,9 β -epoxide, and thence into the chlorohydrin (8). In conjunction with the results¹ already obtained, therefore, the present work provides a complete total synthesis of precalciferol₃.

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¹ J. Dixon, P. S. Littlewood, B. Lythgoe, and A. K. Saksena, *Chem. Comm.*, 1970, 993.

² H. H. Inhoffen, S. Schütz, P. Rossberg, O. Berges, K.-H. Nordsiek, H. Plenio, and E. Höroldt, *Chem. Ber.*, 1958, **91**, 2626; H. H. Inhoffen, H. Burkhardt, and G. Quinkert, *ibid.*, 1959, **92**, 1564; H. H. Inhoffen, G. Friedrich, D. Kampe, and O. Berges, *ibid.*, p. 1772.

³ For nomenclature, see L. F. Fieser and M. Fieser, "Steroids," Reinhold, New York, 1959, p. 338.

⁴ W. S. Johnson, L. Werthemann, W. R. Bartlett, T. J. Brocksom, T. Li, D. J. Faulkner, and M. R. Petersen, *J. Amer. Chem. Soc.*, 1970, **92**, 741.

⁵ I. J. Bolton, R. G. Harrison, and B. Lythgoe, preceding communication.

⁶ C. Herschmann, *Helv. Chim. Acta*, 1949, **32**, 2537.

⁷ I. J. Bolton, R. G. Harrison, and B. Lythgoe, to be published.

⁸ D. H. Williams, Ph.D. Thesis, Leeds, 1961.

⁹ B. Lythgoe and A. K. Saksena, to be published.